

Thermodynamic and practical aspects of oxyfuel CO₂ capture strategies for kraft recovery boilers

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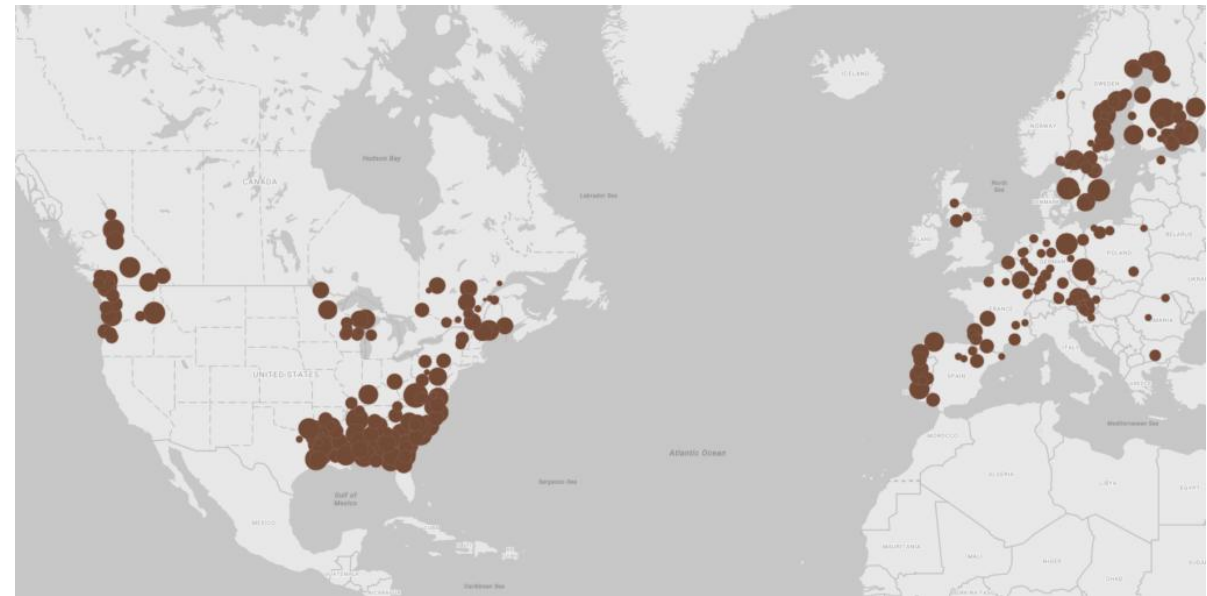
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Introduction – Pulp mills as BECCS

- Pulp and paper industries (PPI) represent a source of biogenic CO₂ emissions
 - Ca. 92 Mt-CO₂/y in Europe [1], 101 Mt-CO₂/y in USA [2]
- Sequestering biogenic CO₂ from PPI can provide net-negative emissions as a BECCS system.
 - ~150-1200 Gt-CO₂ forecast for BECCS removal by 2100 [3]
 - ~2.5-20 Gt/y over 2040-2100



Biogenic CO₂ emissions from PPI sites [4]

[1] Lipiäinen et al. (2023)

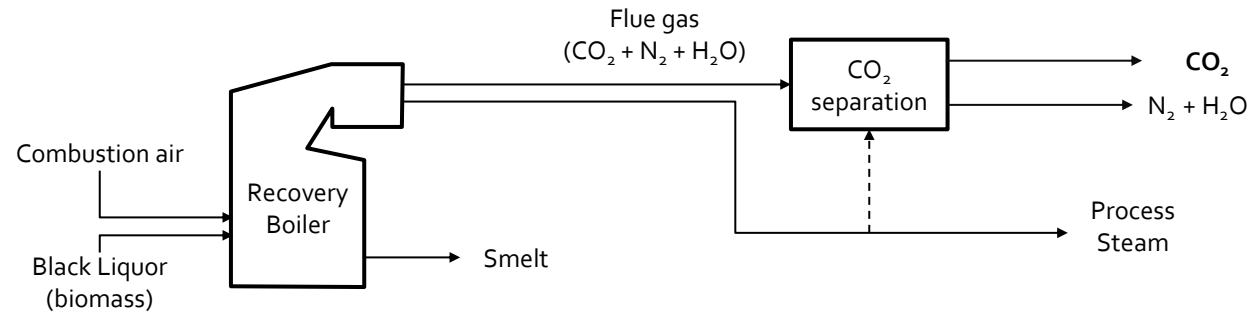
[2] US EPA (2022)

[3] IPCC (2018)

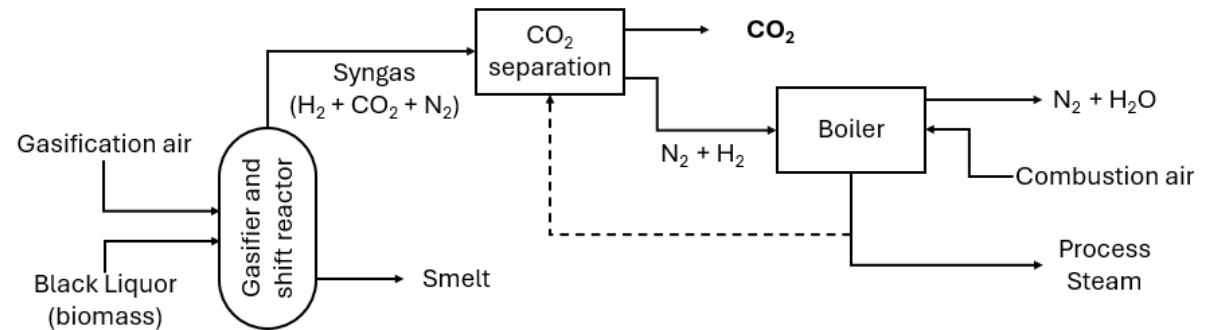
[4] Endrava (2024)

Introduction – CCS processes

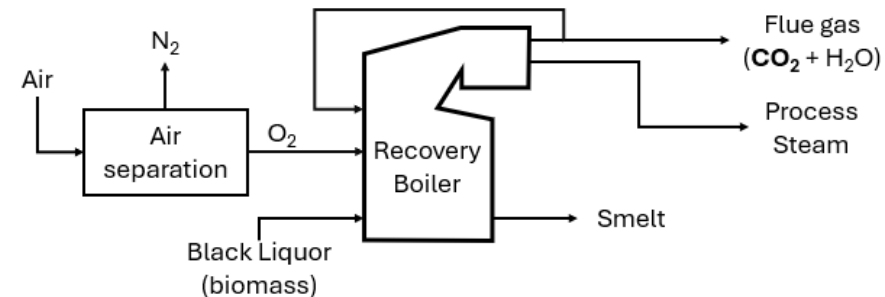
- Post-Combustion



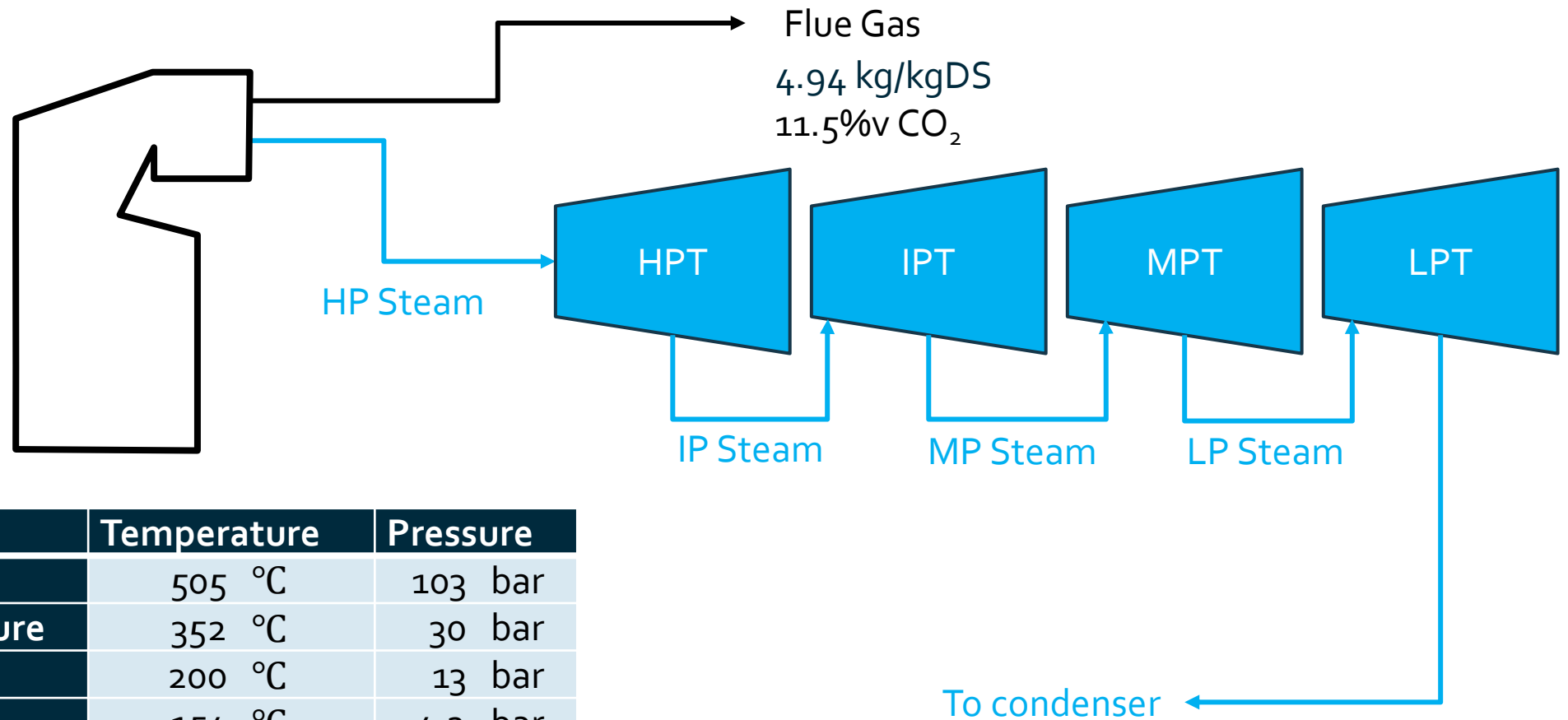
- Pre-Combustion



- Oxyfuel Combustion

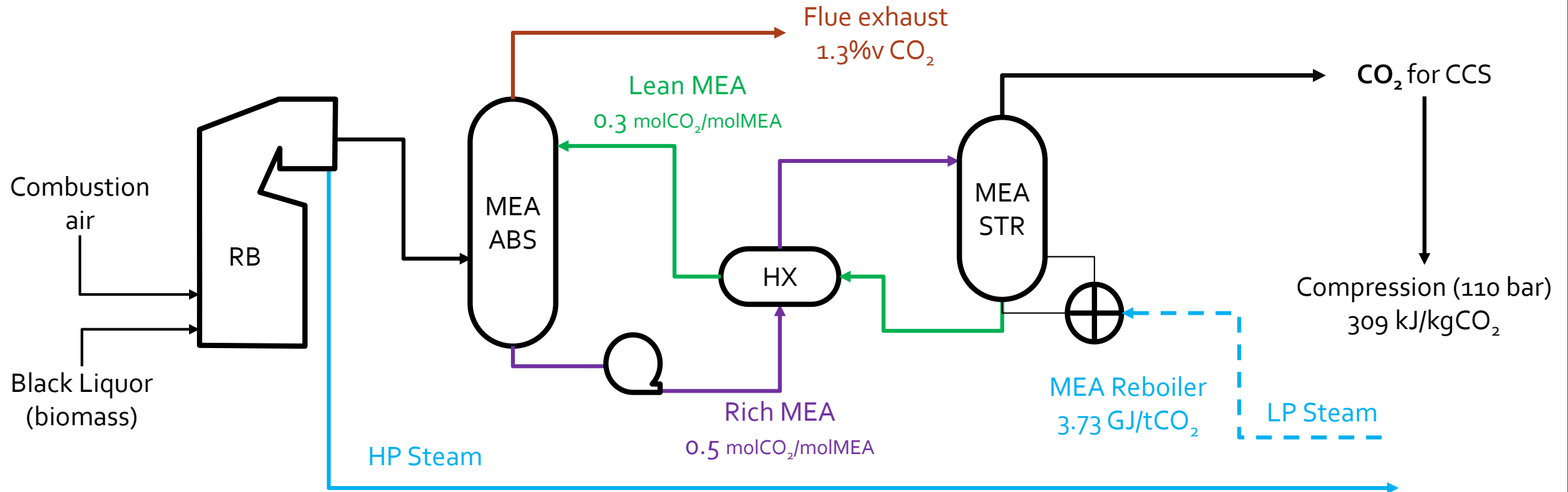


Reference Recovery Boiler

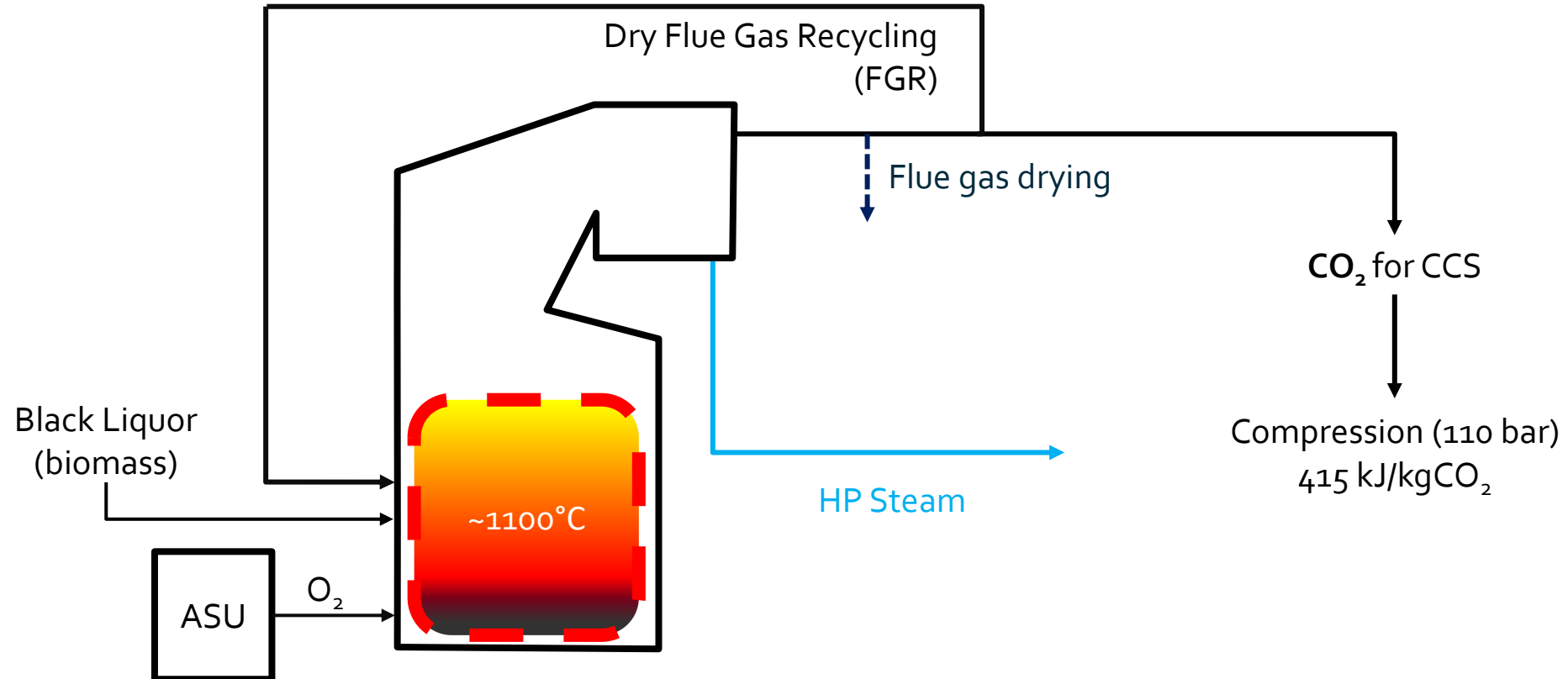


Steam Properties	Temperature	Pressure
High pressure	505 °C	103 bar
Intermediate pressure	352 °C	30 bar
Medium pressure	200 °C	13 bar
Low pressure	154 °C	4.2 bar

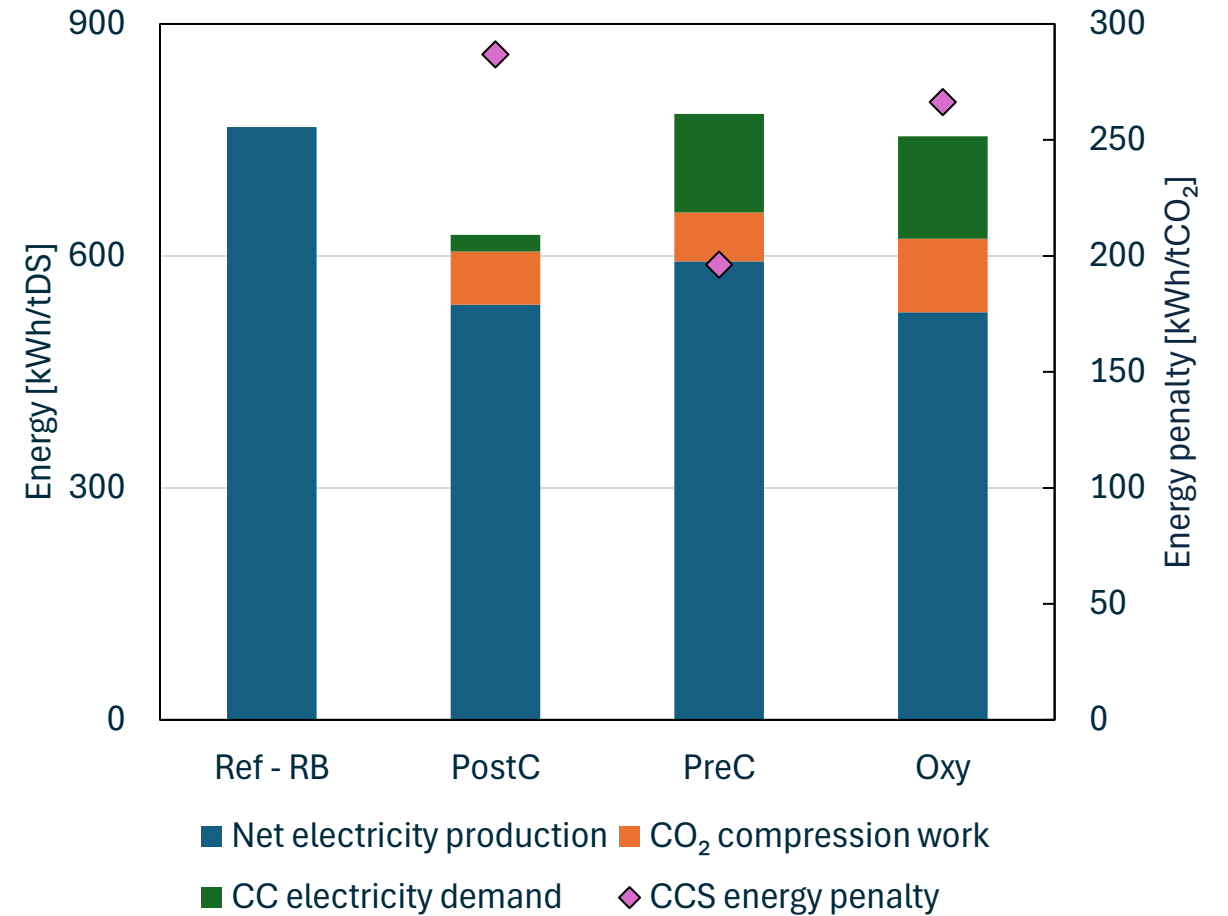
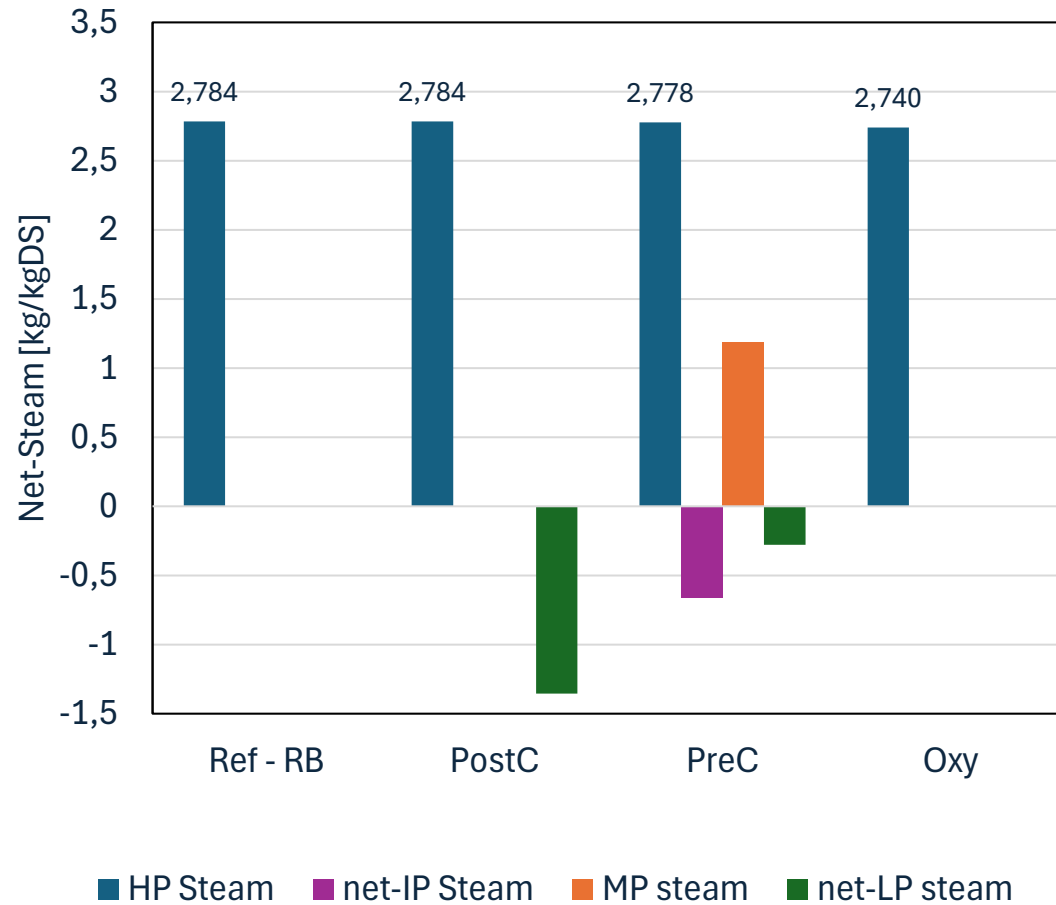
Post-combustion (PostC)



Oxyfuel combustion (Oxy)



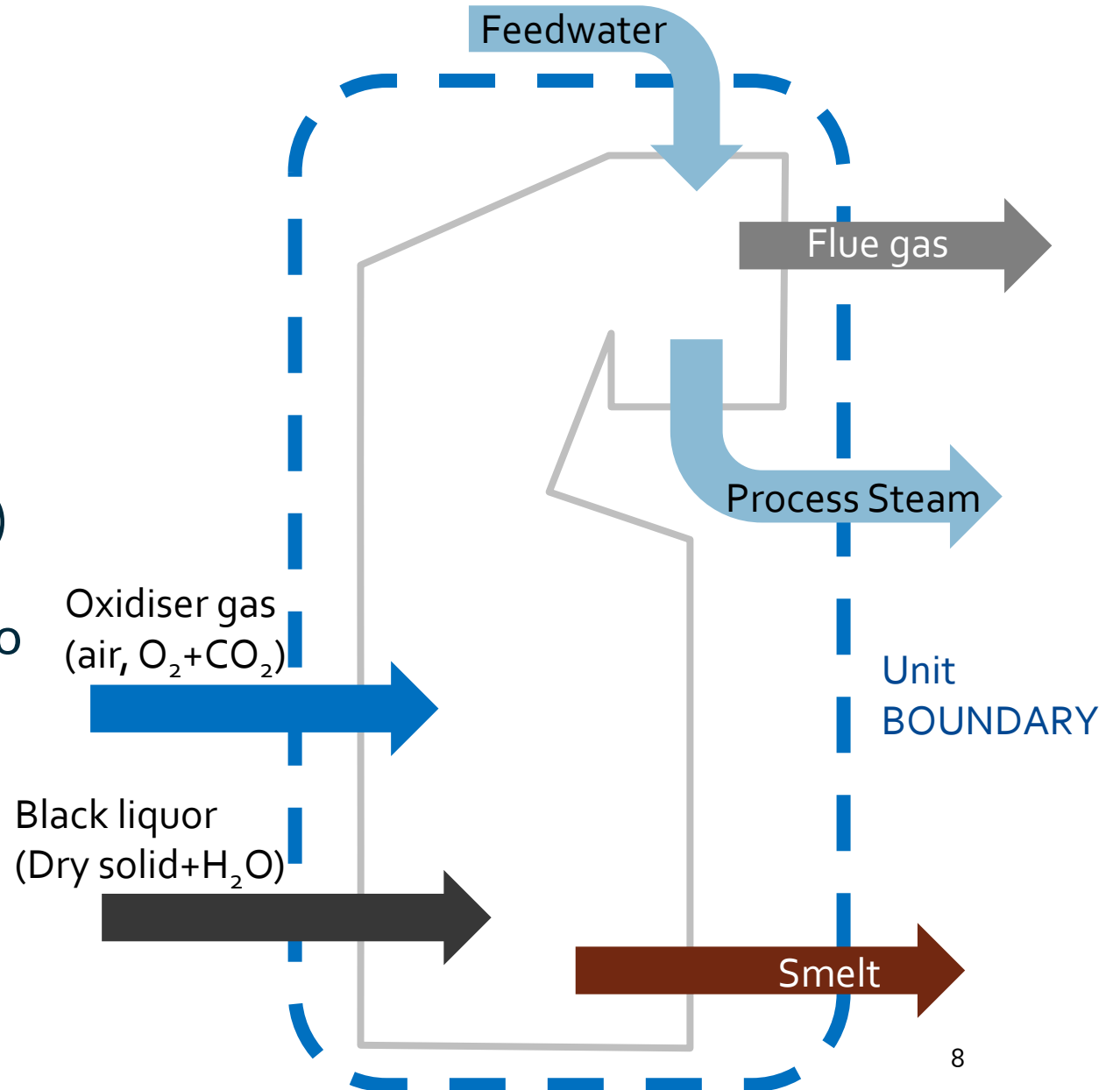
CCS processes



[5] Greencorn et al. (2025)

Methodology

- Aligned with TAPPI recovery boiler performance calculation (TIP 0416-01)
- Thermodynamic data from NASA polynomials (NASA/TP–2002-211556)
- Steam properties from Moran & Shapiro (2006)



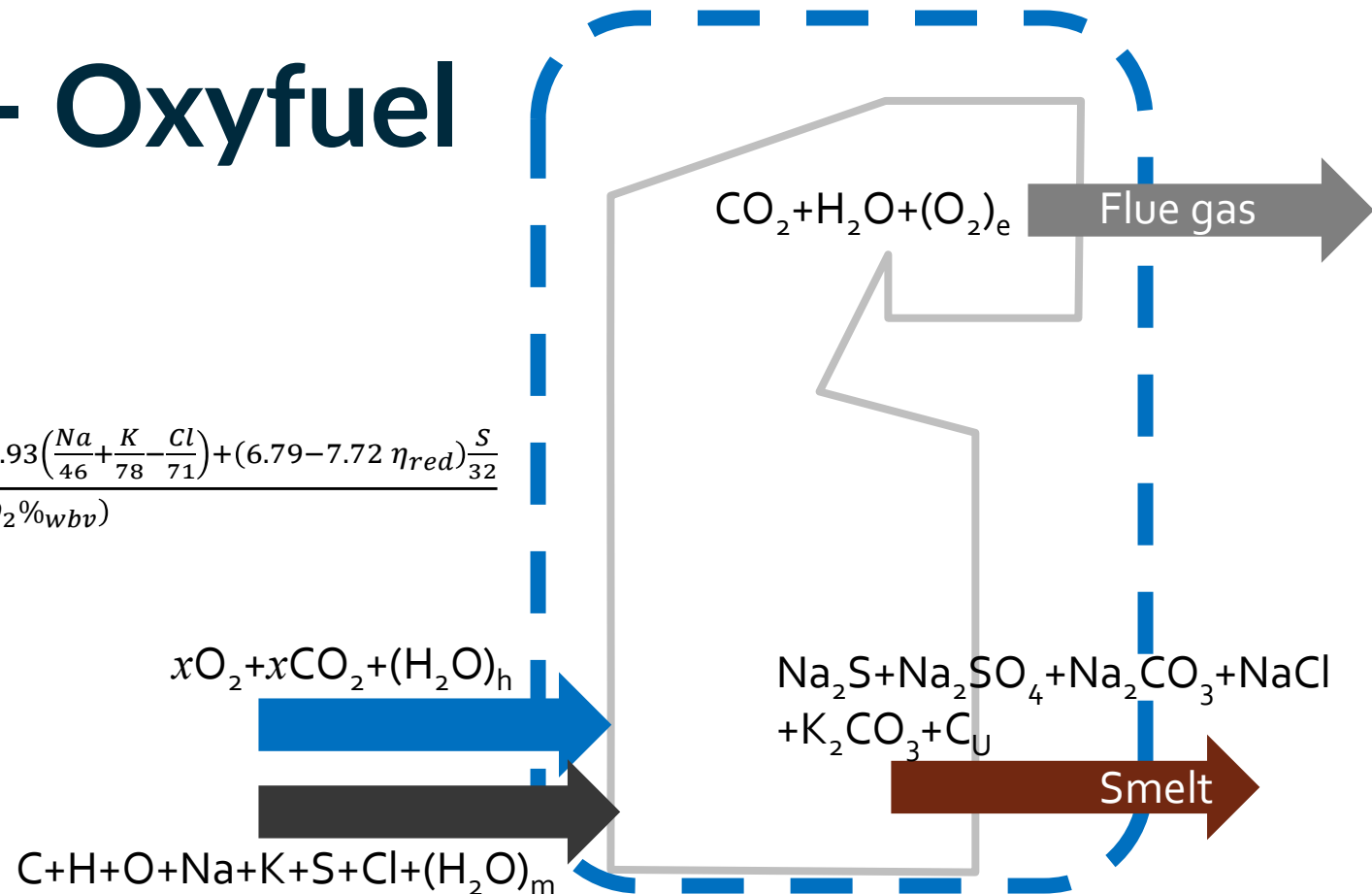
Methodology - Oxyfuel

- Modification of TAPPI calculation
 - flue gas moles:

$$\text{TAPPI calculation: } (FG)_w = \frac{\frac{4.86}{12}C + \frac{2.93}{2}H + \left(\frac{H_2O(m+h)}{18}\right) - \frac{3.88}{32}O + 0.93\left(\frac{Na}{46} + \frac{K}{78} - \frac{Cl}{71}\right) + (6.79 - 7.72 \eta_{red})\frac{S}{32}}{(1 - 4.8 O_2\%_{wbv})}$$

- Mass balance derives new calculation for flue gas moles:

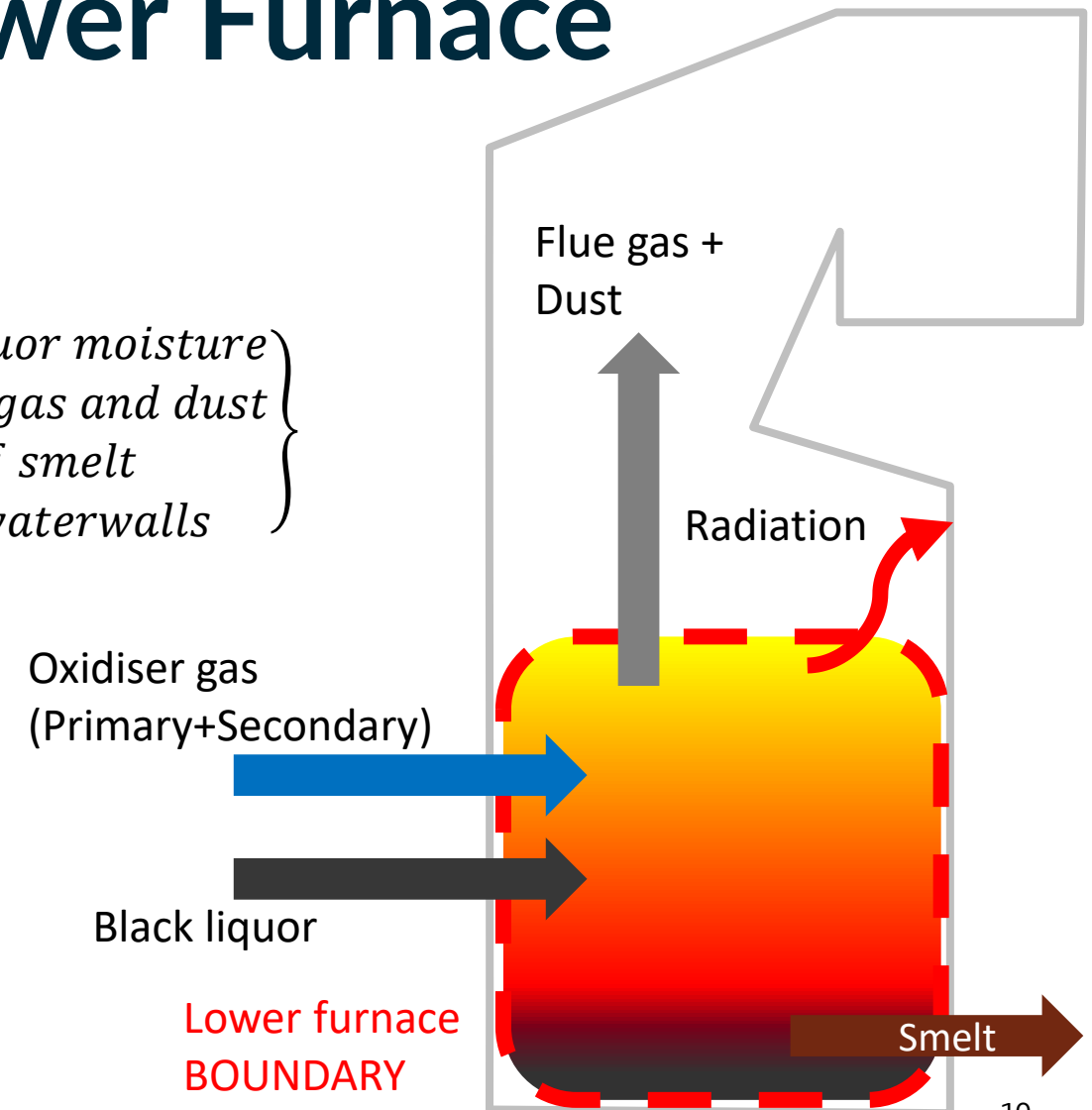
$$(FG)_w = \frac{\left(1 + \frac{x_{CO_2}}{x_{O_2}}\right)\frac{C - C_u}{12} + \left(1 + \frac{x_{CO_2}}{2x_{O_2}}\right)\frac{H}{2} + \left(\frac{H_2O(m+h)}{18}\right) - \frac{x_{CO_2}}{x_{O_2}} * \frac{O}{32} + \left(\frac{x_{CO_2}}{2x_{O_2}} - 1\right)\left(\frac{Na}{46} + \frac{K}{78} - \frac{Cl}{71}\right) + \left(1 + \frac{x_{CO_2}}{x_{O_2}}\left(\frac{3}{2} - 2\eta_{red}\right)\right)\frac{S}{32}}{\left(1 - \left(\frac{x_{CO_2}}{x_{O_2}} + 1\right) * O_2\%_{wbv}\right)}$$



Methodology – Lower Furnace

- Lower furnace energy balance (Adams, 1997)

$$\left\{ \begin{array}{l} \text{Chemical energy from} \\ \text{black liquor combustion} \\ + \text{Sensible energy of} \\ \text{black liquor and oxidizer} \end{array} \right\} = \left\{ \begin{array}{l} \text{Latent heat of black liquor moisture} \\ + \text{Sensible heat of flue gas and dust} \\ + \text{Sensible heat of smelt} \\ + \text{Radiation heat to waterwalls} \end{array} \right\}$$



Methodology – Lower Furnace

- Chemical energy from partial combustion = $\dot{m}_{DS}(LHV)F_h$
- Sensible energy of black liquor and oxidizer
= $(\dot{m}_{DS} + \dot{m}_w + \dot{m}_{SC})C_{P,liq}(T_{liq,i} - T_{ref}) + \dot{m}_{ox}C_{P,ox}(T_{ox,i} - T_{ref})$
- Latent heat of black liquor moisture = $\dot{m}_w\Delta h_v$
- Sensible heat of flue gas and dust = $(\dot{m}_{FG} + \dot{m}_{dust})C_{P,FG}(T_F - T_{ref})$
- Sensible heat of smelt = $\dot{m}_{smelt}C_{P,smelt}(T_{smelt} - T_{ref})$
- Radiation to waterwalls = $(\varepsilon_g 4WL_F + \{1 - 0.6 \varepsilon_g\}W^2)\sigma^\circ(T_F - T_{ref})$

Air Reference Case - Inputs

- BLDS Heating value (HHV=13.2 MJ/kg)
- BLDS concentration (81.3 wt%DS)
- Excess O₂ in flue gas (3.0 v%)
- Furnace Prim+Secd air (65%)
- BL flow (~5300 tDS/d)
 - Floor loading (3.61 MW/m²)
- HBL inlet temperature (140°C)
- Smelt temperature (950°C)
- Air inlet temperature (150°C)
- Flue gas exit temperature (180°C)
- Process steam conditions (515°C, 110 bar)

- BLDS composition (%w):

Carbon	32.20%
Hydrogen	3.30%
Oxygen	33.91%
Sodium	21.40%
Potassium	2.40%
Sulfur	6.40%
Chloride	0.30%
Nitrogen	0.09%

Reference Boiler

- Steam: 2.782 kg/kgDS
- Flue gas: 5.119 kg/kgDS
183.3 mol/kgDS

Composition:

12.55%v CO₂

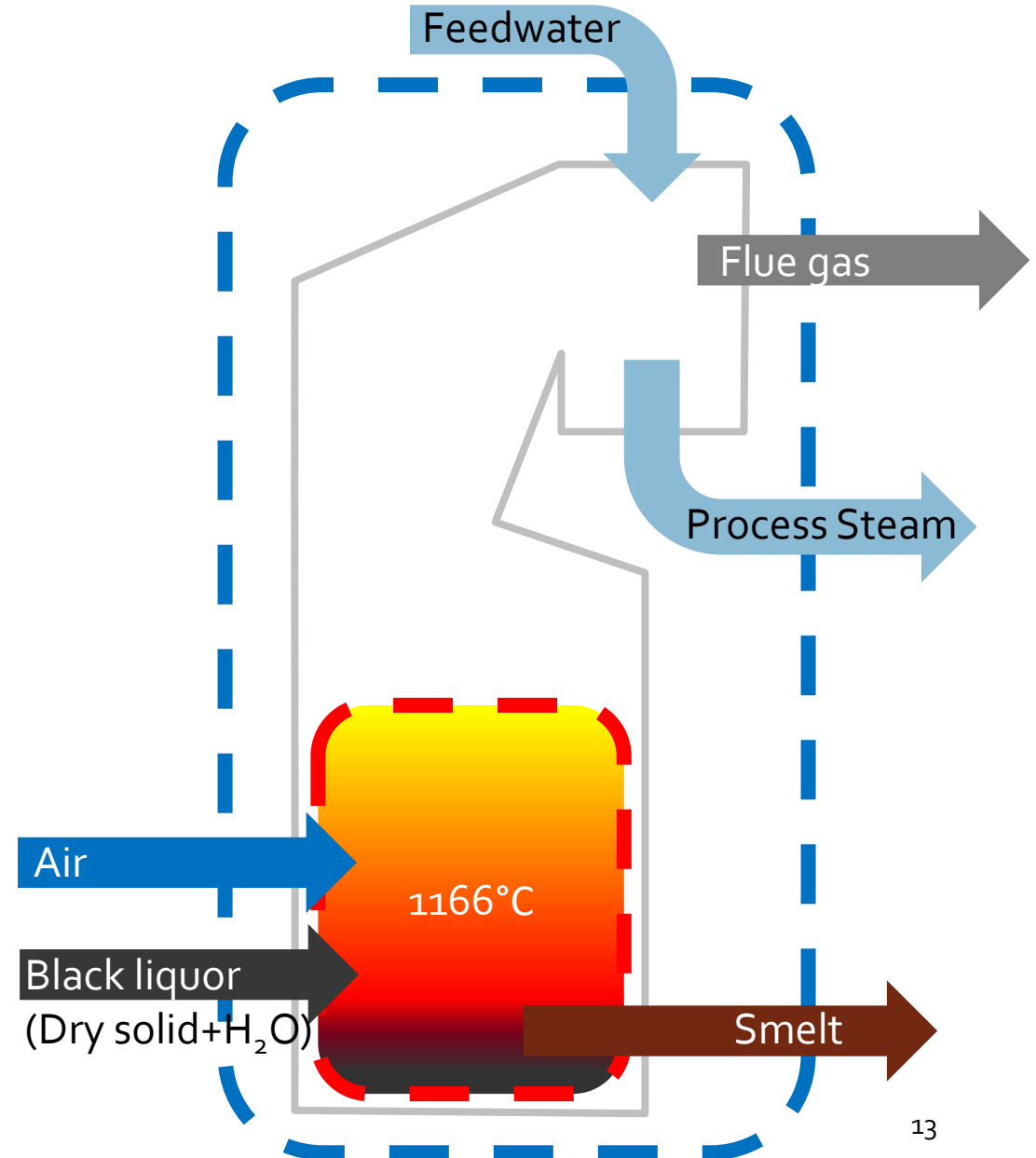
62.28%v N₂

0.020%v CO

0.007%v SO₂

3.00%v O₂

22.14%v H₂O



Oxyfuel Boiler

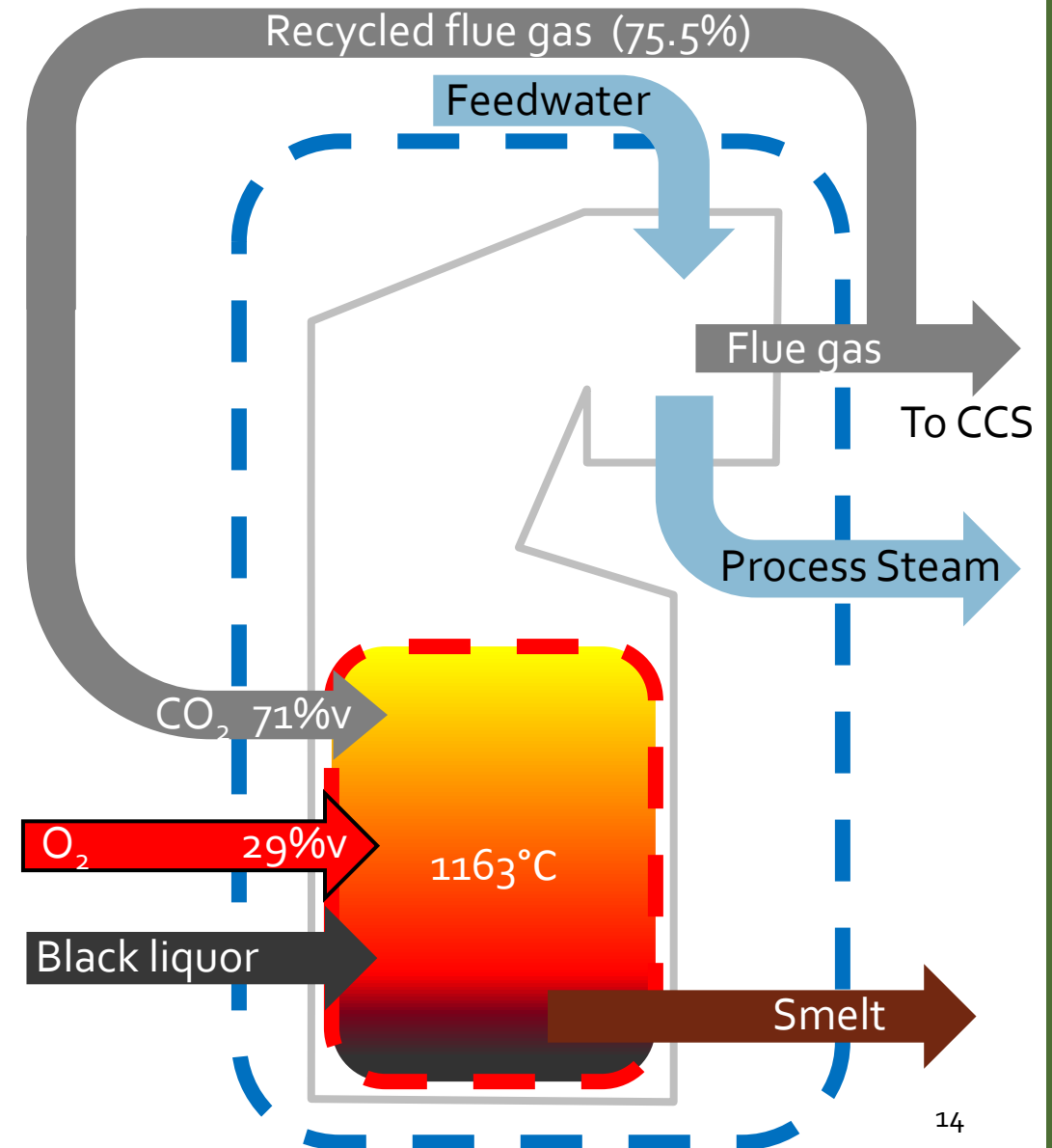
- Steam: 2.782 kg/kgDS
- Flue gas: 5.000 kg/kgDS
138.9 mol/kgDS
- O₂ from ASU:
0.828 kg/kgDS
- To CCS (dry):
1.045 kg/kgDS

Composition:

67.76%v CO₂
0.023%v N₂
0.026%v CO
0.010%v SO₂
2.98%v O₂
29.20%v H₂O

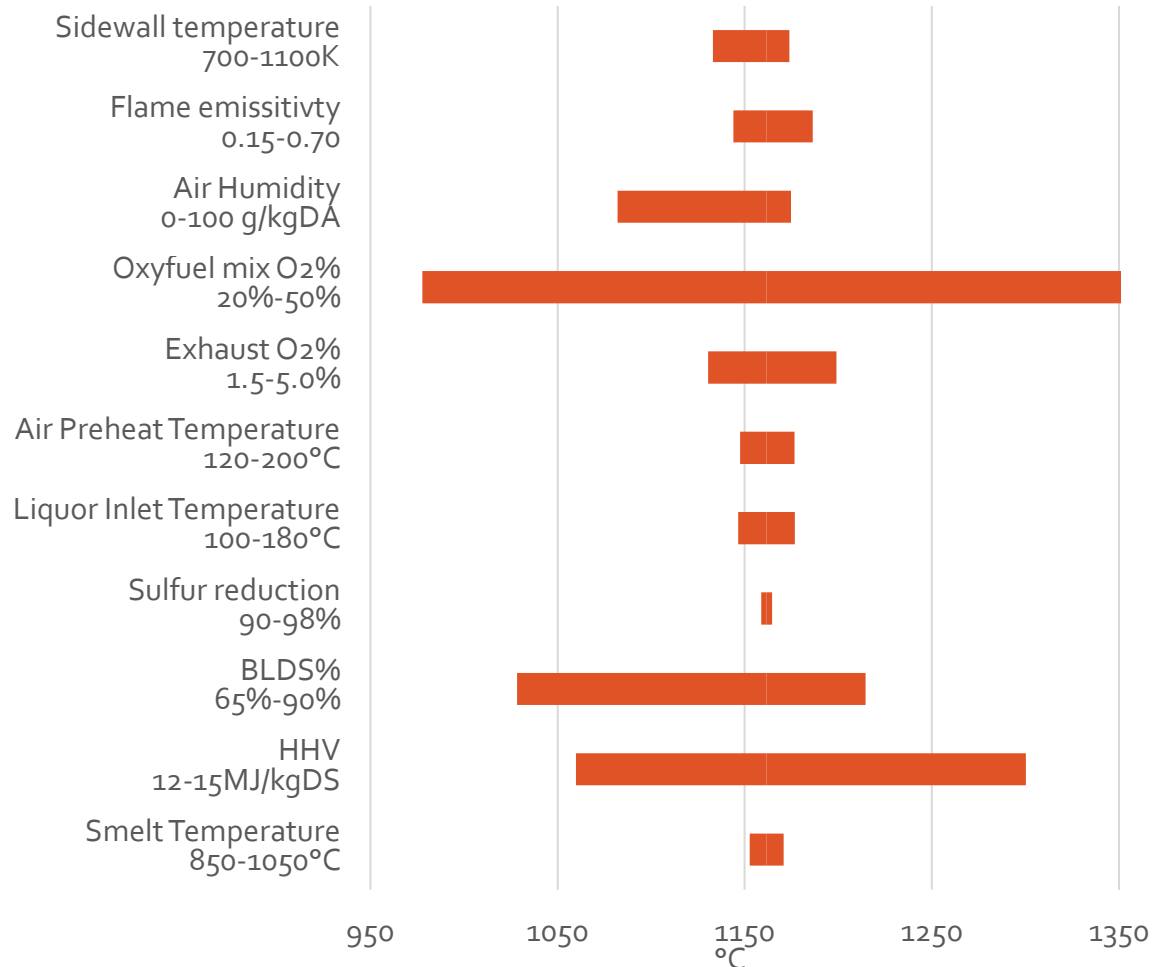
Composition:

95.71%v CO₂
0.033%v N₂
0.037%v CO
0.014%v SO₂
4.21%v O₂

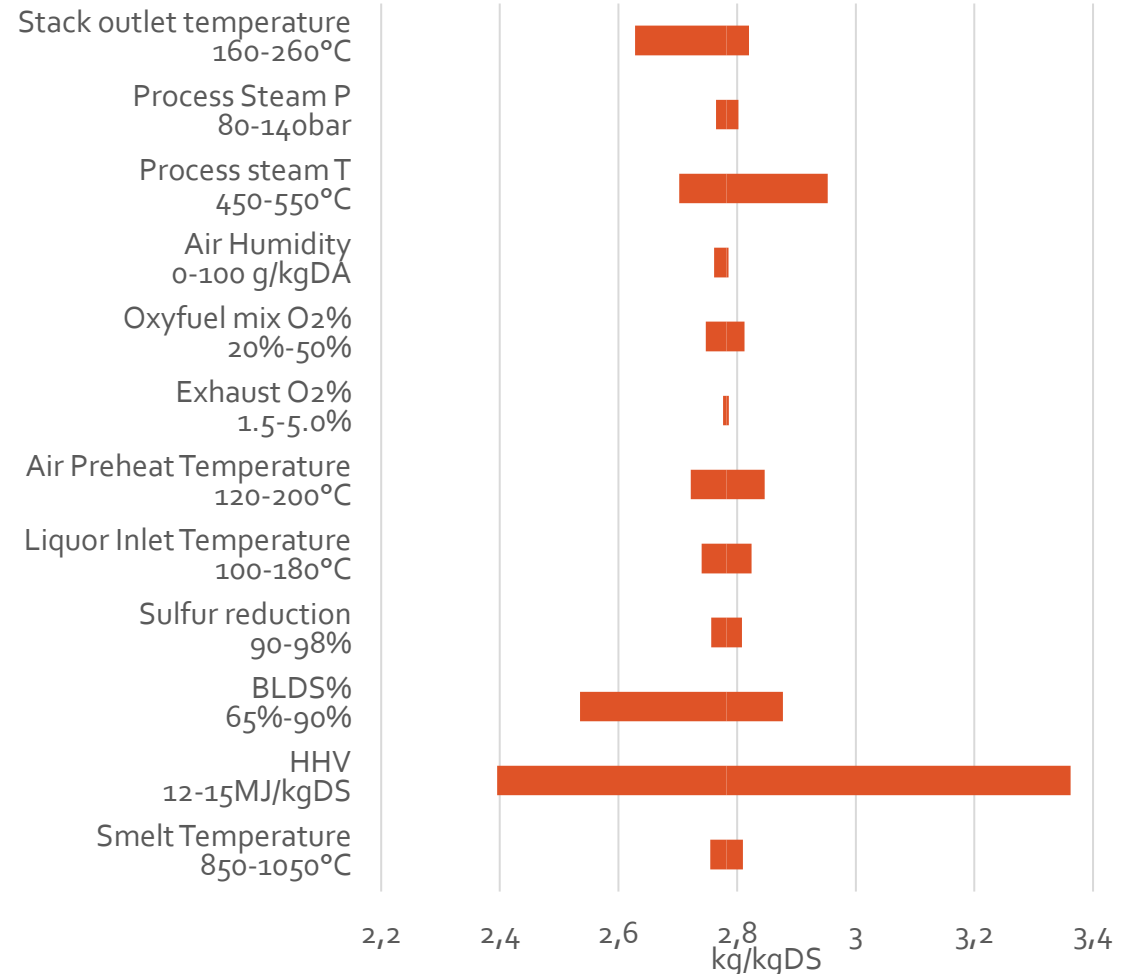


Oxyfuel Sensitivities

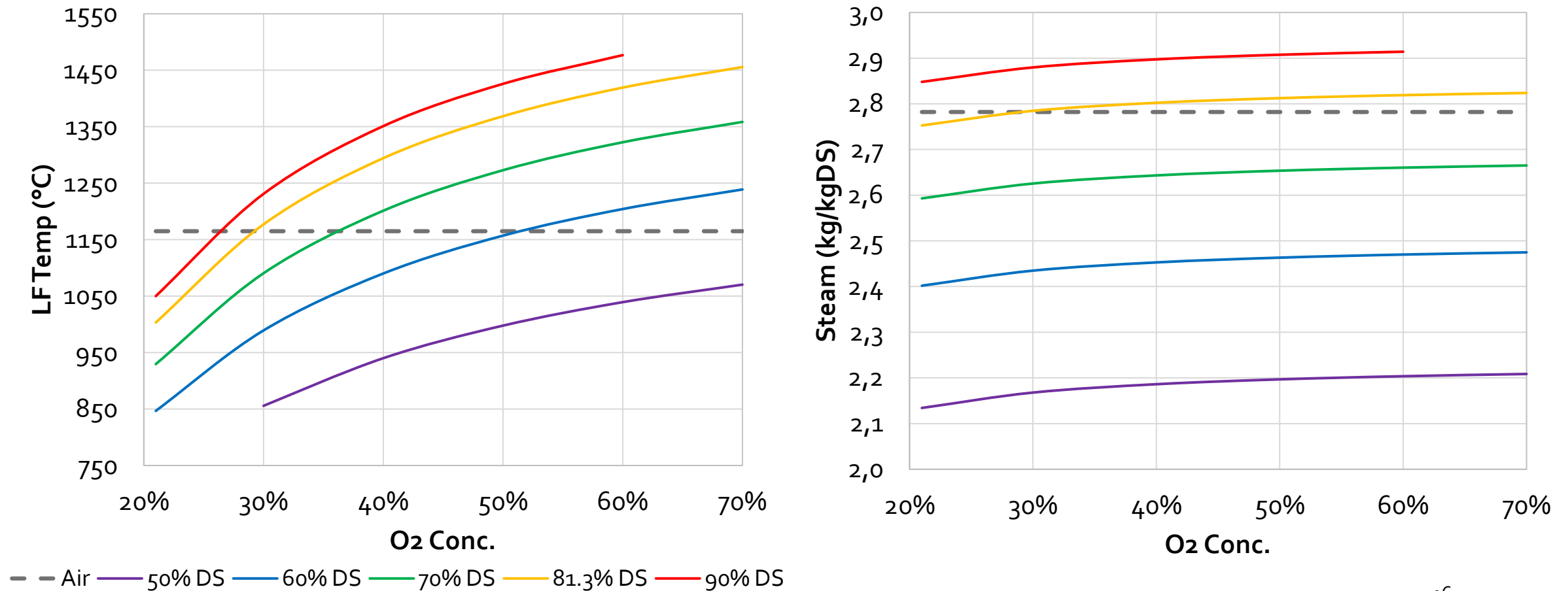
Lower Furnace Temperature



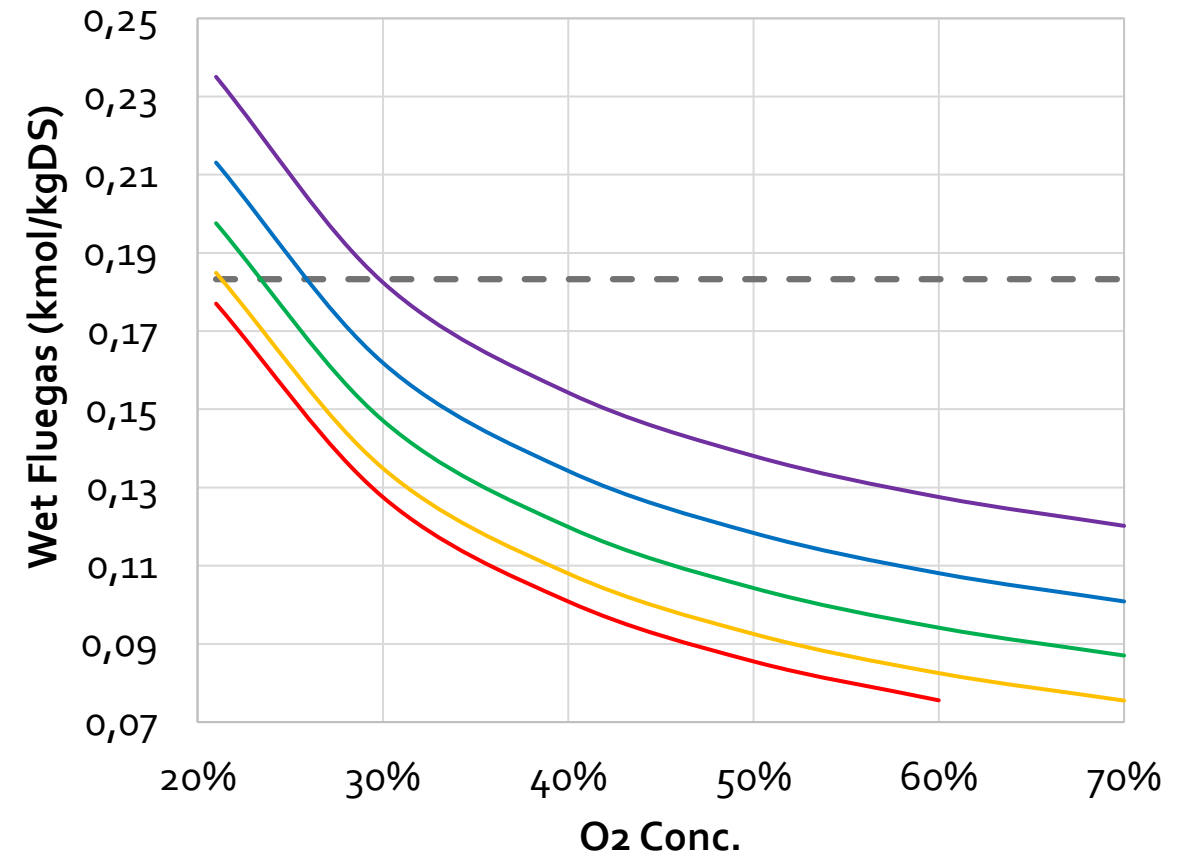
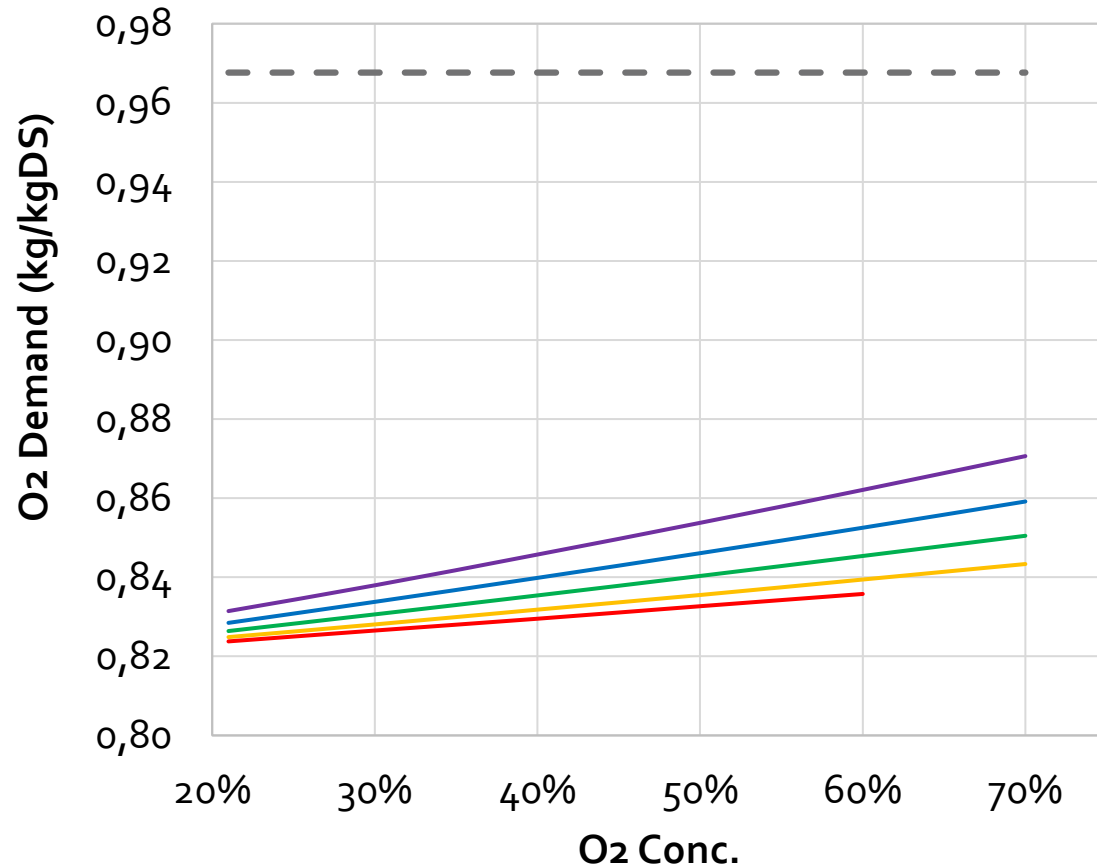
Steam Production



Oxyfuel Variations



Oxyfuel Variations



— Air — 50% DS — 60% DS — 70% DS — 81.3% DS — 90% DS

Oxyfuel gas flows

- **Alternate case:** equivalent molar flows in boiler
 - Increase BLDS inputs: 6760 tpd (+27%)
 - Floor loading: 4.59 MW/m²
- Steam: 2.779 kg/kgDS (+46.13 kg/s)
- Flue gas: 5.238 kg/kgDS (+94.6 kg/s)
144.2 mol/kgDS

Composition:

68.75%v CO₂

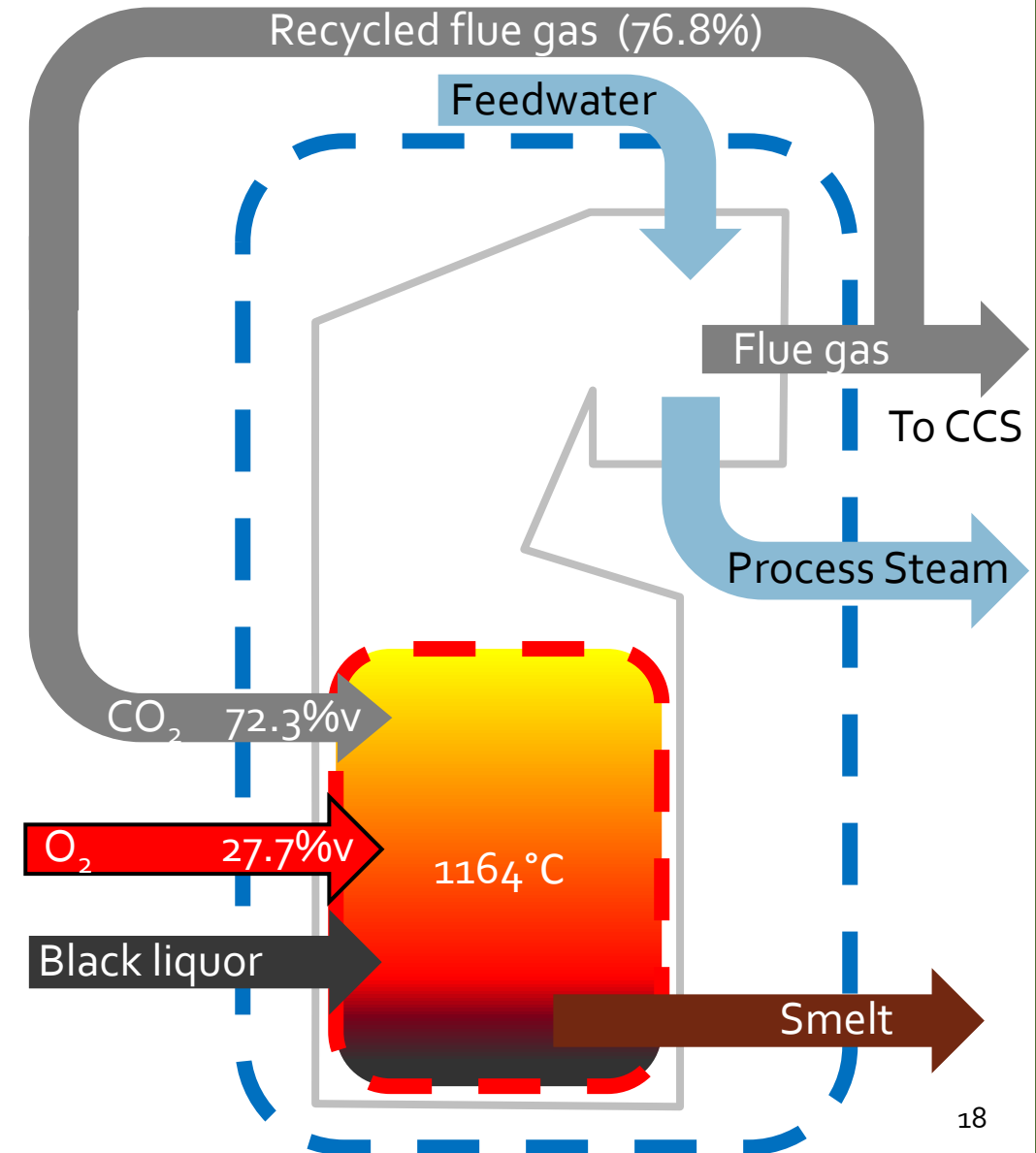
0.022%v N₂

0.025%v CO

0.009%v SO₂

3.00%v O₂

28.20%v H₂O

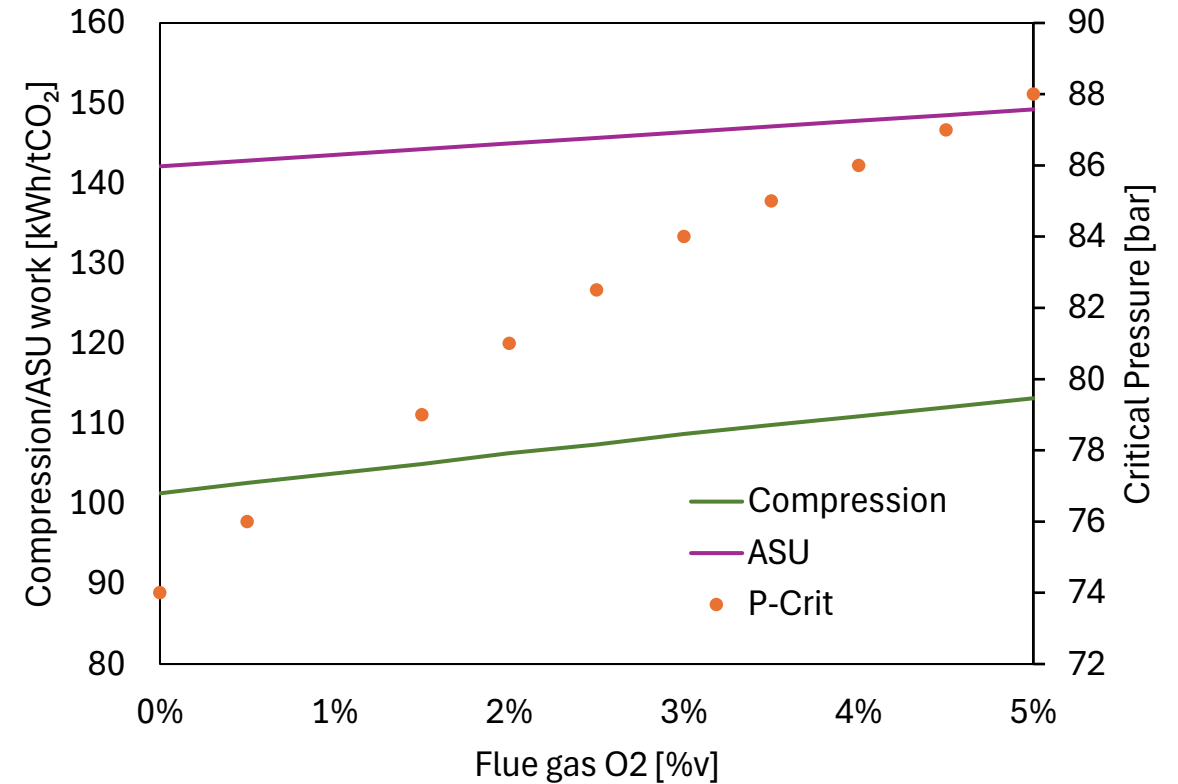
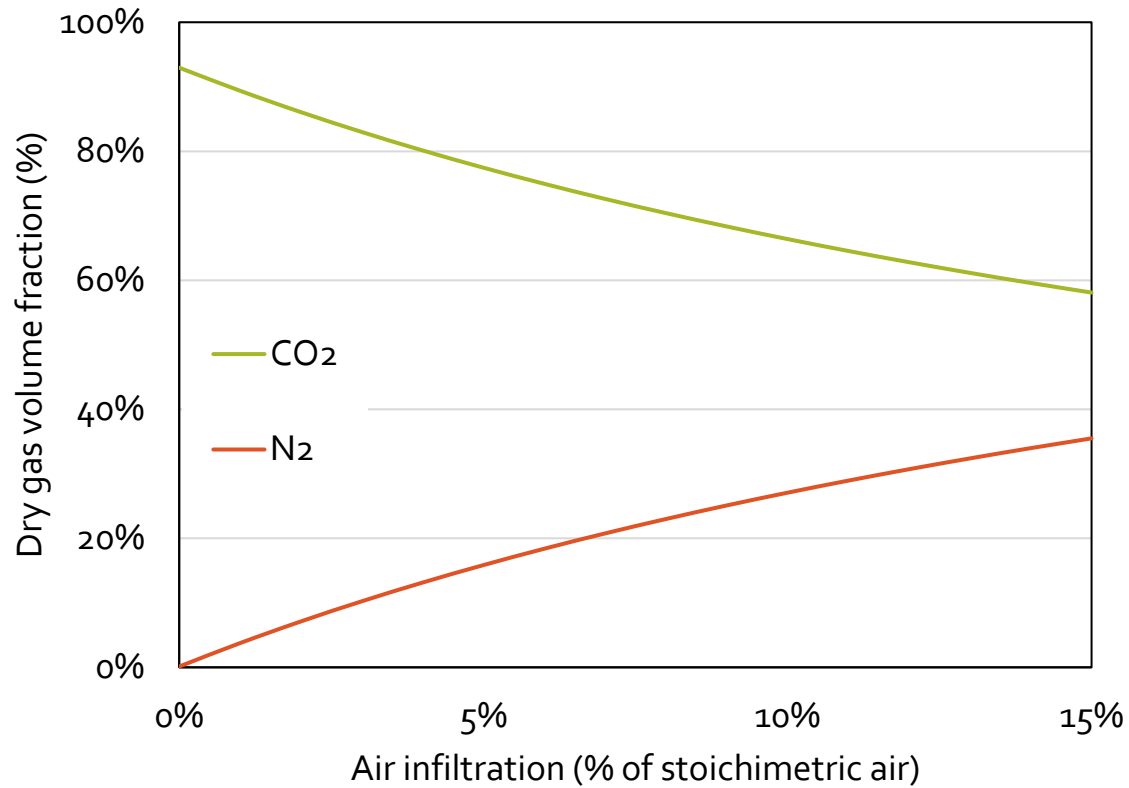


CO₂ quality

Gas quality requirements for selected CCS storage projects

	DYNAMIS (deVisser, 2008)	Porthos (2022)	Northern Lights (2024)	NETL (2019)	Oxy-Kraft Boiler
CO ₂	>95.5%	>95%	>99.81%	>95%	95.705%
H ₂ O	500 ppm	<70 ppm	<30 ppm	500 ppm	-
O ₂	-	<40 ppm	<10 ppm	< 1000 ppm	4.212 %
CH ₄	<4%	<1%	<100 ppm	<4%	-
CO	2000 ppm	<750 ppm	<100 ppm	<35 ppm	370 ppm
N ₂	<4%	<2.4%	<50 ppm	<4%	330 ppm
Ar	<4%	<0.4%	<100 ppm	<4%	-
H ₂	<4%	<0.75%	<50 ppm	<4%	-
H ₂ S	200 ppm	<5 ppm	<9 ppm	<0.01%	-
SO ₂	-	<20 ppm	<10 ppm	<100 ppm	140 ppm

Air infiltration and excess O₂



Conclusions

- Lower furnace Oxyfuel conditions (temperature) are influenced by dry flue gas recycling and BL dry solids concentration
 - Control of these parameters can be used to set lower furnace temperature
 - Higher amounts of FGR and BL water content reduce specific steam production in the boiler
- Oxyfuel conditions reduce volume flows through the boiler
 - Increase firing rate/floor loading possible to match volume flows
- Flue gas quality may be unsuitable for direct sequestration – additional purification required
 - Minimizing contaminants, infiltration air, and excess O₂ reduces energy requirements of purification
- Effect of Oxyfuel conditions on black liquor combustion and smelt recovery is uncertain
 - Parallel research on this topic ongoing within the Oxy-Kraft project



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